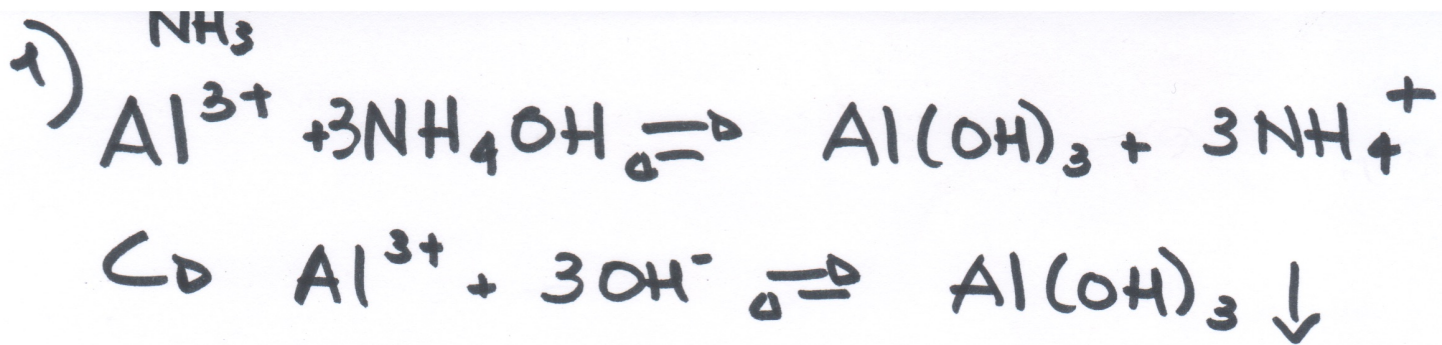
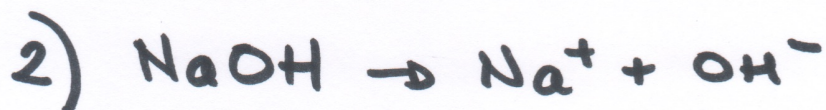
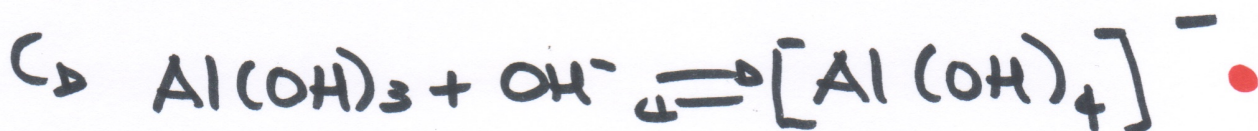


$\rightarrow$ TAMPONE $\text{NH}_4^+ / \text{NH}_3$

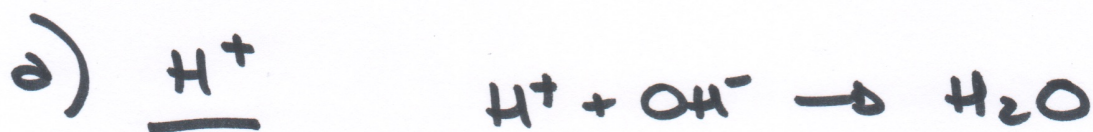
CONC. DI OH^- ADEGUATA PER UNA
 PRECIPITAZIONE COMPLETA



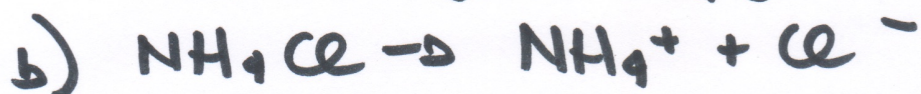
in eccesso di OH^-



$\times$ OTTENERE NUOVAMENTE $\text{Al}(\text{OH})_3 \downarrow$



DIMINUISCE $[\text{OH}^-]$ E IL PRECEDENTE
 EQUILIBRIO SI SPOSTA VERSO SINISTRA



TAMPONE $\text{NH}_3 / \text{NH}_4^+$ pH ~ 9

[OH⁻] NON È SUFFICIENTE PER FORMARE
[Al(OH)₄]⁻

oppure

