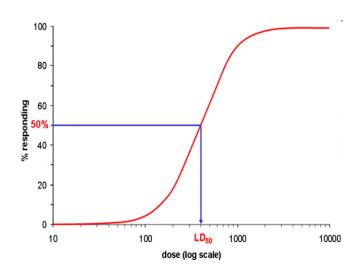
Toxicology

Toxicology studies adverse effects of chemicals on living organisms.

Chemicals exert toxic effects by disturbing the normal course of biochemical reactions. Molecular mechanism.

Quantitative relationship:

- Dose-effect
- LD₅₀



General toxic effects

✓ Nonspecific effects: lipophilic and amphipathic compounds interact with cell membranes. Aliphatic alcohols: antiseptic. Depression of CNS /heart muscle.

✓ Lipid peroxidation: radical compounds oxidate unsaturated fatty acid derivatives. (rancid fat reaction). Detected with thiobarbituric acid → fluorescent product with malondialdehde and similar compounds

Figure 7.9 Radical-induced peroxidation of unsaturated fatty acids.

General toxic effects

✓ Lipid peroxidation:

- 1) damage to membranes
- 2) damage to membrane-bound enzyme systems
- 3) diffusion to the nucleus and DNA-damage

✓ Acidosis:

pH lowering. Blood 7.35-7.40. Effect of ${\rm CO_2}$ (lung, kidneys). Enzyme carboanhydrase catalyzes :

$$2 H_2O + CO_2 \rightarrow H_2CO_3 + H_2O (HCO_3^- + H_3O^+)$$

Decreasing pH \rightarrow \rightarrow increasing breathing frequency

Acidosis by high quantity of acid drugs: salycilic derivatives; formic acid precursors; carboxylic halides; anhydrides; metabolic carboxylic acids (oxalic acid from ethylene glycol)

General toxic effects

✓ Oxygen deficiency: stops energy accumulation in the cell (ATP, GTP) and oxidative phosphorilation (respiratory chain)

Carbon monoxide: 0.1% in air blocks 50% of a human blood Hb

oxyhemoglobin (Fe²⁺)

carboxyhemoglobin



General toxic effects

✓ Oxygen deficiency: stops energy accumulation in the cell (ATP, GTP) and oxidative phosphorilation (respiratory chain)

Methemoglobinemia: balanced by NADH-methemoglobin reductase

General toxic effects

✓ ATP deficiency:

a) inhibition of the citric acid cicle

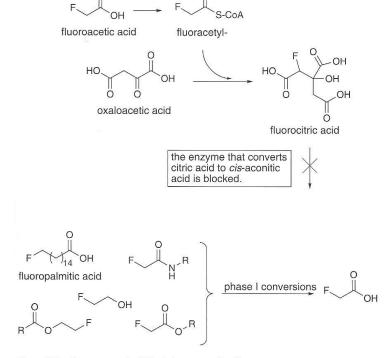


Figure 7.20 Fluoracetic acid will block the citric acid cycle.

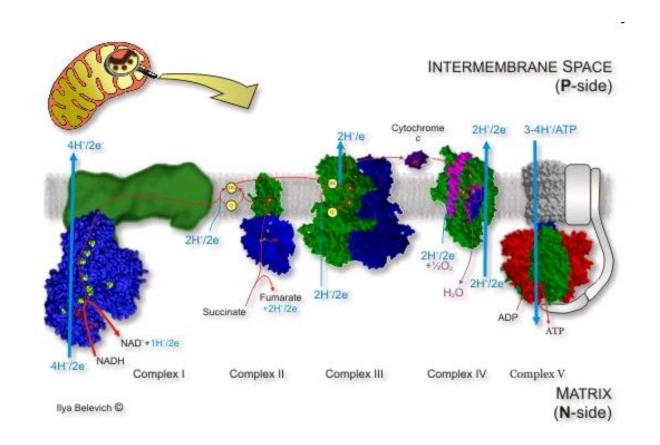


Dichapetalum cimosum (CH₂F-COOH)

General toxic effects

✓ ATP deficiency:

b) inhibition of the respiratory chain



General toxic effects

✓ ATP deficiency:

Hydrogen cyanide Hydrogen sulphide

b) inhibition of the respiratory chain

Figure 7.22 Enzymatic hydrolysis of cyanogenic glucosides.

Specific toxic effects

- ✓ Effects on the blood: O_2 / oxyhemoglobin. Hypoxia (CO-Hb, metHb). Aplastic anemia: reduction of blood cell formation. Leukemia: leukocytes cancer.
- ✓ Effects on the cardiovascular system: Bradycardia, tachycardia, arrhythmia (heart).

 Arteriosclerosis / atherosclerosis (vases). Acrolein, As.

Specific toxic effects

- ✓ Effects on the nervous system:
 - interaction with <u>receptors and ion channels</u> (1. Agonistic or antagonistic effect on receptors;
 - 2. Agonistic effect on receptors 3. Antagonistic effect on receptors; 4) neurotransmitter metabolic enzymatic inhibition (Acetylcholine esterase, AchE*); 5) ion channels blocking)
 - inhibition of <u>ATP production</u>;

Specific toxic effects

✓ Effects on the nervous system: AchE: inhibitors

Organophosphate military poisons:

nerve gases; irreversible AchE inhibitors. Among the most toxic synthetic compounds ever made (World War II)

✓ Sarin: isopropyl methylphosphonofluoridate. Systemic poison to CNS. One drop can kill a human. LD_{50} 0.01 mg/Kg; mp -58°C, bp 147°C.

- ፟፟ጷ <u>Tokyo 1995,</u> terroristic attack to the subway by オウム真理教 (Aum Shinrikyo, new religious movement")
- ✓ **Tabun:** ethyl dimethylphosphoramidocyanidate. LD₅₀ (mouse, i.p.) 0.6 mg/Kg; mp -49°C, bp 238°C.
- ✓ VX: O-Ethyl S-[2-(Diisopropylamino)ethyl]methyl-phosphonothioate

Specific toxic effects ✓ *Effects on the nervous system*: ion channels blockers

Name	Structure	Formula and MW	action
Tetrodotoxin (TTX)	H ₂ N [®] HO OH OH OH	319 (C ₁₁ H ₁₇ N ₃ O ₈)	Blocks sodium channels LD ₅₀ 10 μg/Kg
Saxitoxin (STX)	H ₂ N O H OH OH OH	301 (C ₁₀ H ₁₉ N ₇ O ₄) ²⁺	Blocks sodium channels
Neosaxitoxin (NeoSTX)	HO N H NH @ NH2 NH OH OH	317 (C ₁₀ H ₁₉ N ₇ O ₅) ²⁺	Blocks sodium channels
Gonyautoxin (GTX)	$\begin{array}{c ccccccccccccccccccccccccccccccccccc$	I, IV: 380; II, III: 364	Blocks sodium channels
Tetramine (TTM)	CH ₃ CH ₃ —N – CH ₃ CH ₃ CH ₃	74 (C ₄ H ₁₂ N)	Ganglionic blocking agent (curare like)





Specific toxic effects

✓ Effects on the liver: fatty liver (radicals, CCl₄) / liver cancer.
Induction of cytochrome P450
(vinyl chloride → epoxide)

✓ Effects on the kidneys: acidosis, oxalic acid, metals (Hg, Cd)

Figure 7.26 Compounds toxic to the liver.

Specific toxic effects

✓ Effects on the respiratory system: lung cancer (smoke). Paraquat radical. Pyrrolizidine alkaloids.

✓ Effects on the skin: Local effects of strongly reactive compounds: chemical burns.

Photosensitization.



Figure 7.27 Compounds that are toxic to the lung.

Figure 7.28 Phototoxic and photoallergic compounds.

Specific toxic effects

✓ Effects on the immune system:

Physiological function: protection against microorganisms, viruses, malignant cells. Troublesome hyper-reaction of immune system: <u>allergy</u>:

- non specific allergic reaction: phagocytosys, inflammation
- specific allergic reaction: determination of self/foreign nature with antibody (by B-cells) and lymphocytes (by T-cells) production. Antigenes: MW > 5000 Da. Haptene: small molecule + protein

<u>Sensitization</u>: generation of long-lived memory cells that enable tho body to remember that an antigen is foreign. Quick reaction (minutes vs. 12-48 h of normal immune reaction).

Specific toxic effects

✓ **Chemical allergenes:** *immediate hypersensitivity*: antibody mediated asthma and contact dermatitis.

Anaphylactic shock. Cross-reactivity: different diisocyanates.

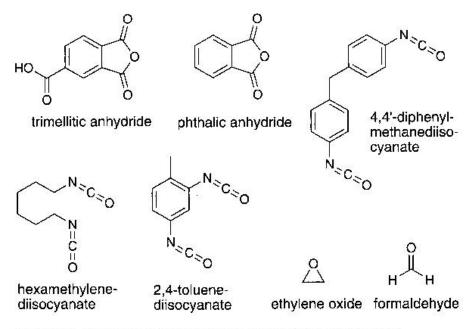


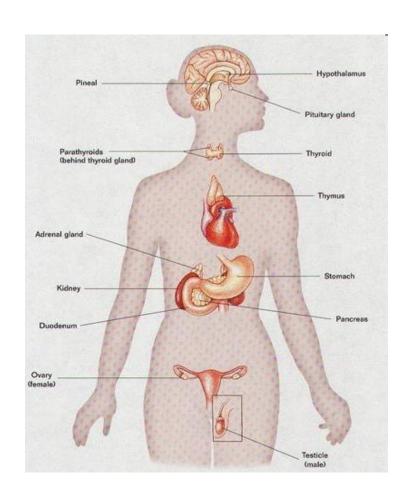
Figure 7.30 Examples of chemicals that give immediate hypersensitivity.

Specific toxic effects

✓ Endocrine disruptors: effects on endocrine system: regulation of blood pressure regulation of body temperature reproduction building of muscle mass and bones

Toxicity: agonists / antagonists

Environmental toxicants



Specific toxic effects

✓ Endocrine disruptors: estrogenic compounds

Environmental Science and Technology 2009, **43**, 597. SPE; GC-MS; LC-MS.

		source (n = 19)		
	MRL	max.	med.	10.
known or potential EDCs				
atrazine	0.25	870	32	15
17β-estradiol	0.50	17	17	1
estrone	0.20	0.90	0.30	15
17 a.ethynylestradiol	1.0	1.4	1.4	1
bisphenol A	5.0	14	6.1	3
butylbenzyl phthalate	50	54	53	2
diethylhexyl phthalate	120	170	150	2
galaxolide	25	48	3	4
linuron	0.50	9.3	4.1	5
nonylphenol	80	130	100	8
progesterone	0.50	3.1	2.2	4
testosterone	0.50	1.2	1.1	2



DDT

Specific toxic effects

✓ **Chemical teratogenes:** *Effects on the embryo*: first 8 weeks gestation (histogenesis). Influence the following organogenesis (weeks 3-9). There are also physical and biological t. More than 1000 chemicals are teratogens on rodents (pH of rodent embryo blood > maternal: trapping of weak acids). Approx. 30 are teratogen for humans. Hg (Minamata), Pb, Cd, As; thalidomide (day 23-38 from conception, 10000 cases in the 60s)

Figure 7.32 Examples of teratogenic compounds.

thalidomide

Genotoxic effects

DNA as molecular target:

- ✓ chemical origin. Known from 1775, Dr Percivall Pott: scrotal cancer was the first occupational (chimney sweeps) link to cancer.
 - → environmental toxicology
 - → epidemiology





Genotoxic effects

DNA bases modification:

- ✓ with exogenous chemicals
 - electrophiles
 - radicals (or reactive products formed from)
 - metals

DNA-Nu:
$$\stackrel{}{\longrightarrow}$$
 E+ $\stackrel{}{\longrightarrow}$ DNA-Nu⁺-E

DNA-Base

DNA-H $\stackrel{}{\longrightarrow}$ R $\stackrel{}{\longrightarrow}$ DNA- + R-H

t½ guanine 15000000 h t½ 7-alkylguanine 200 h

Genotoxic effects

DNA bases modification:

• bifunctional electrophiles

Figure 8.6 Intra- and interstrand cross links by the bifunctional epichlorohydrin, which to the left has reacted with adenine and cytosine in the same strand and to the right with guanine and thymine in opposite strands.

Genotoxic effects

DNA bases modification:

• radicals

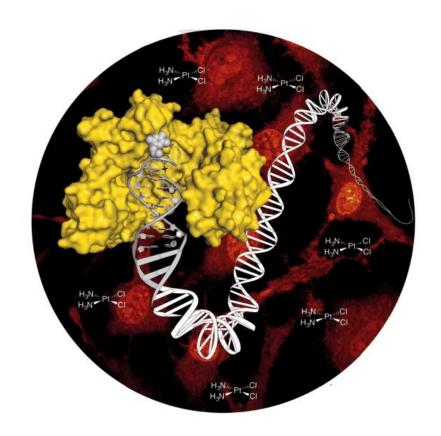
Figure 7.4 The antitumor antibiotic calicheamicin γ_i^{Br} and its interaction with DNA.

both strands cutting \rightarrow chromosomal aberration

Genotoxic effects

DNA bases modification:

- metal ions (Ni²⁺, Cd²⁺, Cr⁶⁺)
- metal complexes (Pt²⁺)

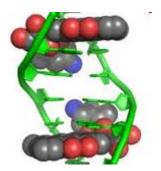


Angewandte Chemie International Edition 2010, 49, 2949

Genotoxic effects

Intercalation:

- √ flat molecules
 - PAH
 - aflatoxins
 - daunomycin



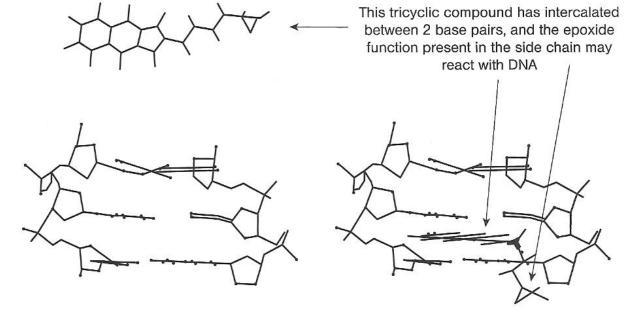


Figure 8.7 Intercalation.

Genotoxic effects

Intercalation:

✓ flat molecules: **PAHs**: Formed by incomplete combustion of other hydrocarbons. Condensed ring system with high C/H ratio. Sources include engine exhaust, wood stove smoke, cigarette smoke, charbroiled food, tires, creosote. Poorly water soluble: soil accumulation.

Pyrene nomenclature

Figure 9.1 Examples of relatively common and dangerous PAHs.

7,8-Diol-9,10-epoxide, carcinogenic (+)anti-isomer

OH 7,8-Diol

Genotoxic effects

Microtubules damage:

- ✓ Microtubules are protein structures that move chromosomes during mitosis
 - colchicine